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$p_{\text{atm}} = (20.32 \text{ psia}) (14.70 \text{ psia}) =$   
 $5.62 \text{ psi}$   $p_{\text{gage}} = 5.62 \text{ psig c.}$

Situation: Pressure values need to be converted. Find: Calculate the absolute pressure (psia)

corresponding to a pressure of 0.55 bar (gage). Properties:  $p_{\text{atm}} =$

$14.70 \text{ psi}$ : Solution:  $p_{\text{gage}} = 0.55 \text{ bar}$  1

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