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state and fixed
composition;
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B. Liquid-
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Boyle's law. For a given mass of gas at constant temperature, the volume of gas varies inversely with the pressure.

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Both: $\text{Sn}^{2+} + (\text{E} \circ \text{red} = -0.14\text{V}; \text{E} \circ \text{ox} = -0.15\text{V})$, $\text{Fe}^{2+} + (\text{E} \circ \text{red} = -0.44\text{V}, \text{E} \circ \text{ox} = -0.77\text{V})$, $\text{Cl}_2 + (\text{E} \circ \text{red} = 1.36\text{V}, \text{E} \circ \text{ox} = -1.47\text{V})$.
Oxidizing agents

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(list in order of decreasing E° red): $\text{Cl}_2 > \text{H}^+ > \text{Sn}^{2+} > \text{Fe}^{2+}$. Reducing agents (list in order of decreasing E° ox): $\text{Mg} > \text{Ni} > \text{Sn}^{2+} > \text{Fe}^{2+} > \text{Cl}_2 > \text{Au}$. 4.

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